

BITSAT 2025 May 30 Shift 2 Question Paper

Time Allowed :3 Hours	Maximum Marks :390	Total questions :130
------------------------------	---------------------------	-----------------------------

General Instructions

Read the following instructions very carefully and strictly follow them:

1. **Exam Mode:** Computer Based Test
2. **BITSAT exam duration:** 3 hours
3. **Medium of Exam:** English
4. **BITSAT exam Sections:**
 - Part I - Physics (30 questions)
 - Part II - Chemistry (30 questions)
 - Part III - English Proficiency (10 questions) and Logical Reasoning (20 questions)
 - Part IV - Mathematics/Biology (40 questions)
5. **Type of Questions:** Multiple Choice Questions (MCQ)
6. **BITSAT Total Questions:** 130 Questions
7. **BITSAT Exam Pattern Total Marks:** 390 Marks

1. If $x + \frac{1}{x} = 4$, find the value of $x^4 + \frac{1}{x^4}$.

- (A) 194
 - (B) 1945
 - (C) 190
 - (D) 1940
-

2. The equation of the line passing through the point (2, 3) and making equal intercepts on the coordinate axes is:

- (A) $x + y = 5$
 - (B) $3x + 2y = 12$
 - (C) $2x + 3y = 12$
 - (D) $5x + 5y = 25$
-

3. If $\tan \theta + \cot \theta = 4$, then find the value of $\tan^3 \theta + \cot^3 \theta$.

- (A) 52
 - (B) 44
 - (C) 46
 - (D) 54
-

4. If $y = \ln(x^2 + 1)$, then find $\frac{dy}{dx}$ at $x = 1$.

- A) $\frac{1}{2}$
 - B) $\frac{1}{3}$
 - C) 1
 - D) $\frac{2}{3}$
-

5. If a, b are roots of the equation $x^2 - 5x + 6 = 0$, find the value of $a^3 + b^3$.

- (A) 125
 - (B) 215
 - (C) 98
 - (D) 35
-

6. In triangle ABC , the length of sides are $AB = 7$, $BC = 10$, and $AC = 5$. What is the length of the median drawn from vertex B ?

- (A) 6
 - (B) 5
 - (C) 7
 - (D) 8
-

7. If $f(x) = e^{2x} \sin x$, find $f'(x)$.

- (A) $e^{2x}(2 \sin x + \cos x)$
 - (B) $e^{2x}(2 \sin x - \cos x)$
 - (C) $e^{2x}(2 \cos x + \sin x)$
 - (D) $e^{2x}(\sin x - 2 \cos x)$
-

8. The half-life of a radioactive substance is 4 hours. If initially there are 256 grams, how much remains after 10 hours?

- (A) 45.26 g
 - (B) 16 g
 - (C) 64 g
 - (D) 8 g
-

9. A fluid flows through a pipe with varying cross-section. If the velocity at the narrow section is 3 m/s and the cross-sectional area is half of the wider section, what is the velocity in the wider section?

- (A) 1.5 m/s
 - (B) 6 m/s
 - (C) 0.5 m/s
 - (D) 3 m/s
-

10. The escape velocity from the surface of a planet is 11.2 km/s. If the radius of the planet is doubled but the mass remains the same, what will be the new escape velocity?

- (A) 22.4 km/s

- (B) 7.9 km/s
(C) 15.8 km/s
(D) 5.6 km/s
-

11. Two identical charges q are placed 1 m apart. The electrostatic force between them is 9×10^{-9} N. What is the magnitude of each charge? (Take $k = 9 \times 10^9 \text{ Nm}^2/\text{C}^2$)

- (A) 1×10^{-9} C
(B) 3×10^{-9} C
(C) 1×10^{-9} C
(D) 3×10^{-9} C
-

12. A charged particle moves in a magnetic field with velocity v perpendicular to the field B . The radius of the circular path is r . Which of the following expressions gives the charge q of the particle?

- A) $q = \frac{mv}{Br}$
B) $q = \frac{mvB}{r}$
C) $q = \frac{mB}{vr}$
D) $q = \frac{Bvr}{m}$
-

13. How much water must be added to 500 mL of 2 M H_2SO_4 solution to make it 0.5 M?

- (A) 1.5 L
(B) 2.0 L
(C) 1.0 L
(D) 0.75 L
-

Quick Tip

For dilution problems, always use:

$$M_1V_1 = M_2V_2$$

and remember that volume units must be consistent. The volume of water added is the difference between final and initial volumes.

14. Question 14 How many grams of CO_2 are produced when 10 g of C_2H_6 (ethane) is completely combusted? Reaction: $2\text{C}_2\text{H}_6 + 7\text{O}_2 \rightarrow 4\text{CO}_2 + 6\text{H}_2\text{O}$

- (A) 29.3 g
- (B) 44.0 g
- (C) 58.6 g
- (D) 88.0 g

15. A 2 L container holds oxygen gas at 300 K and 2 atm pressure. If the temperature is increased to 600 K and the volume is doubled, what is the final pressure?

- (A) 1 atm
- (B) 2 atm
- (C) 0.5 atm
- (D) 4 atm

16. For the reaction $\text{N}_2 + 3\text{H}_2 \rightleftharpoons 2\text{NH}_3$, if initially 1 mole of N_2 and 3 moles of H_2 are taken and at equilibrium 0.4 moles of NH_3 are formed, find the equilibrium concentration of H_2 .

- (A) 2.4 moles
 - (B) 2.8 moles
 - (C) 3.4 moles
 - (D) 3.0 moles
-